FACULTY OF SCIENCE BSc – IIIyear- Sem V (Practical Examination) Subject: CHEMISTRY Paper –VI (Physical Chemistry) Question Bank

(With effect from 2018-19)

Duration: 3 h

Max. Marks: 25

Question Paper Pattern:

- A. Question for Principle Writing
- **B.** Question for Performing Experiment

A. Questions for the Principle Writing:

Any one among the following may be given:

- 1. Determine the distribution coefficient of iodine between water and carbon tetrachloride/Determine the molecular status and partition coefficient of benzoic acid in toluene and water.
- 2. Determine the distribution coefficient of acetic acid between n-butanol and water.
- 3. Determine the cell constant of a given conductivity cell.
- 4. Determine the dissociation constant (Ka)/Verify Ostwald's dilution law of acetic acid by conductivity measurements.
- **5.** Verify Beer's law for KMnO₄ solution and determination of the concentration of the given solution.
- **6.** Verify Freundlich adsorption isotherm for the adsorption of acetic acid over animal charcoal.
- **7.** Determine the Surface tension of a liquid (density of the liquid, water and surface tension of water are provided).
- **8.** Determine the viscosity of a liquid using Ostwald viscometer (density of the liquid, water and viscosity of water are provided).

B. Question for Performing Experiment:

- 1. Determine the distribution coefficient of iodine between water and carbon tetrachloride/Determine the molecular status and partition coefficient of benzoic acid in toluene and water.
- 2. Determine the distribution coefficient of acetic acid between n-butanol and water.
- 3. Determine the cell constant of a given conductivity cell.
- 4. Determine the dissociation constant (Ka)/Verify Ostwald's dilution law of acetic acid by conductivity measurements.
- 5. Verify Beer's law for KMnO₄ solution and determination of the concentration of the given solution.
- 6. Verify Freundlich adsorption isotherm for the adsorption of acetic acid over animal charcoal.

Scheme of Valuation:

A.	Principle Writing:	05 Marks (Brief principle with necessary equations, model
		graph)
Β.	Experiment:	15 Marks
		Experiment performance with tabulation – 06 Marks
		(a minimum of 5 sets of readings in a non-instrumentation
		experiment or 10 sets of readings in case of instrumentation
		experiment)
		Graph – 05 Marks
		Calculations – 04 Marks
		Result – 01 Mark
C.	Record and Viva:	05 Marks
	TOTAL:	25 Marks